UP CNET Memory Based Paper 21st May 2025 Shift 01

Q1. Which type of immunity is obtained after vaccination?
(a) Innate immunity
(b) Passive immunity
(c) Active acquired immunity
(d) Natural immunity
Ans.(c)

Q2. Which of these is a non-communicable disease?
(a) Influenza
(b) Dengue
(c) Cancer
(d) Tuberculosis
Ans.(c)

Q3. Which of the following diseases is caused by the bacterium Mycobacterium tuberculosis?
(a) Malaria
(b) Tuberculosis
(c) Hepatitis
(d) Typhoid
Ans.(b)

Q4. Malaria is transmitted by:
(a) Housefly
(b) Female Anopheles mosquito
(c) Male Anopheles mosquito
(d) Sandfly
Ans.(b)

Q5. Which vitamin deficiency causes scurvy?
(a) Vitamin A
(b) Vitamin B12
(c) Vitamin C
(d) Vitamin D
Ans.(c)

Q6. A patient with mechanical heart valve on warfarin has INR of 9.0 but no bleeding. What is the immediate management?
(a) Stop warfarin and observe
(b) Give vitamin K orally
(c) Give vitamin K IV and fresh frozen plasma
(d) Give protamine sulfate
Ans.(b)

Q7. Which is the largest internal organ in human body–
(a) Brain
(b) Heart
(c) Kidney
(d) Liver
Ans.(d)

Q8. Which vitamin does convert prothrombin to thrombin?
(a) Vitamin D
(b) Vitamin A
(c) Vitamin K
(d) Vitamin B2
Ans.(c)

Q9. Which one of the following substances is released during the inflammatory process when platelets are activated?
(a) Histamine (Chosen option)
(b) Heparin
(c) Serotonin
(d) Bradykinin
Ans.(c)

Q10. Which of the following vital signs changes is earliest in hypovolemic shock?
(a) Hypotension
(b) Tachycardia
(c) Bradycardia
(d) Narrowed pulse pressure
Ans.(b)

Q11. Which muscle is responsible for extension of the forearm at the elbow joint?
(a) Biceps brachii
(b) Triceps brachii
(c) Brachialis
(d) Deltoid
Ans.(b)

Q12. Portal vein hypertension occurs in –
(a) Diabetes Mellitus
(b) Liver cirrhosis
(c) Cholecystitis
(d) Nephritis
Ans.(b)

Q13. BCG is administered as_____.
(a) Intravenous
(b) Intradermal
(c) Subcutaneous
(d) Intramuscular
Ans.(b)

Q14. What does ECG measure?
(a) Brain activity
(b) Electrical activity of heart
(c) Blood pressure
(d) Lung capacity
Ans.(b)

Q15. Core body temperature is highest at:
(a) Early morning
(b) Noon
(c) Late afternoon
(d) Evening
Ans.(c)

Q16. Inflammation of middle ear is called as_____.
(a) Rhinitis
(b) Mastoiditis
(c) Otitis media
(d) Sinusitis
Ans.(c)

Q17. Which of the following groups of vitamins is fat soluble?
(a) A, B, E, K
(b) A, B, D, K
(c) A, D, E, K
(d) C, D, E, K
Ans.(c)

Q18. The supplementation given along with measles vaccine is _____.
(a) Vitamin C
(b) Vitamin B
(c) Vitamin A
(d) Vitamin D
Ans.(c)

Q19. The source of transmission of cholera is through_____.
(a) Contaminated water and food
(b) Contaminated soil
(c) Poor environmental sanitation
(d) Droplet infection
Ans.(a)

Q20. Elephantiasis is a one of the main clinical feature of____.
(a) Malaria
(b) Filaria
(c) Dengue
(d) Leprosy
Ans.(b)

Q21. "Not providing adequate food or clothing to the child" is an indicator for_____.
(a) Physical abuse
(b) Child neglect
(c) Emotional abuse
(d) Substance abuse
Ans.(b)

Q22. World AIDS day is celebrated on_____ every year.
(a) November 1st
(b) December 1st
(c) March 1st
(d) July 1st
Ans.(b)

Q23. The first breast milk which is rich in nutrients is known as_____.
(a) Foremilk
(b) Colostrums
(c) Hind milk
(d) Artificial milk
Ans.(b)

Q24. Oxytocin injection is administered to a woman after the expulsion of the placenta is to_____.
(a) Relieve pain
(b) Prevent infection
(c) Stimulate uterine contraction
(d) Facilitate rest and sleep
Ans.(c)

Q25. Which of the following elements is a part of the P block of the periodic table?
(a) Rhodium
(b) Hydrogen
(c) Oxygen
(d) Silver
Ans.(c)

Q26. What is the chemical name of the common kitchen ingredient named 'Baking Soda'?
(a) Sodium Bicarbonate
(b) Sodium Hydroxide
(c) Sodium Hypochlorite
(d) Sodium Peroxide
Ans.(a)

Q27. Which of the following options are all transition metals?
(a) F, Cl, Br, I
(b) Cu, Fe, Ni, Ti
(c) He, Ne, Ar, Au
(d) Na, Mg, Al, Mn
Ans.(b)

Q28. Packets of potato chips are usually flushed with which one of following gases to prevent rancidity?
(a) Carbon dioxide
(b) Hydrogen
(c) Nitrogen
(d) Oxygen
Ans.(c)

Q29. Tribromomethane is a synonym for which brominated organic solvent?
(a) Borazine
(b) Diborane
(c) Bromoform
(d) Bromosuccinimide
Ans.(c)

Q30. In industrial settings, Freon-12 is most commonly manufactured by which chemical reaction?
(a) Wurtz reaction
(b) Swarts reaction
(c) Elimination reaction
(d) Substitution reaction
Ans.(b)

Q31. Which of the following statements is correct with reference to the representative elements?
(a) They contain only metals.
(b) They consist of both s- and p-block elements.
(c) They only include p-block elements.
(d) They are also called Transuranium elements.
Ans.(b)

Q32. Which gas is found in soda water?
(a) Hydrogen
(b) Nitrogen
(c) Carbon dioxide
(d) Freon
Ans.(c)

Q33. A salt formed from a strong acid and a weak base will have a pH:
(a) less than 7
(b) equal to 7
(c) greater than 7
(d) Cannot be determined

Ans.(a)

Q34. A proton bonds to the oxygen atom of a solvent water to give a _____ hydronium ion.
(a) square pyramid
(b) trigonal pyramidal
(c) tetrahedral
(d) square planar
Ans.(b)

Q35. The mass of the proton and mass of _____ is the same.
(a) Neutron
(b) Electron
(c) Isoprone
(d) Alpha particle
Ans.(a)

Q36. Which element in the modern periodic table has the same number of valence electrons as that of caesium?
(a) Aluminium
(b) Barium
(c) Potassium
(d) Calcium
Ans.(c)

Q37. In 1866, John Newlands, an English scientist, arranged the then known elements in the order of _____. It is known as 'Newlands' Law of Octaves'.
(a) decreasing atomic masses
(b) increasing atomic number
(c) increasing atomic masses
(d) decreasing atomic number
Ans.(c)

Q38. Out of the following metals, choose the odd one out on the basis of their place in the periodic table.
(a) Sodium
(b) Lithium
(c) Potassium
(d) Magnesium
Ans.(d)

Q39. In 1869, who is credited with publishing the first widely recognized version of the periodic table?
(a) John Alexander Newlands
(b) Dmitri Mendeleev
(c) Johann Döbereiner
(d) Lothar Meyer
Ans.(b)

Q40. Which chemical is used for whitening (or removal of colours) of cloth in cloth industries?
(a) Calcium oxychloride
(b) Sodium hydrogencarbonate
(c) Calcium chloride
(d) Sodium carbonate
Ans.(a)

Q41. Ortho-sulphobenzimide is one of the artificial sweeteners. It is also known as _____.
(a) sucralose
(b) aspartame
(c) saccharin
(d) alitame
Ans.(c)

Q42. Which of the following is NOT used as food preservative?
(a) Table Salt
(b) Sugar
(c) Sodium benzoate
(d) Sodium hydrogen carbonate
Ans.(d)

Q43. Which detergents are also used in toothpaste?
(a) Anionic detergents
(b) Non-ionic detergents
(c) Cationic detergents
(d) Synthetic detergents
Ans.(a)

Q44. Identify the oxidizing agent in the reaction:
$8H^+ + 6NO_2^- + 2MnO_4^- \rightarrow 2Mn^{2+} + 5N_2 + 4H_2O$
(a) MnO_4^-
(b) NO_2^-
(c) H^+
(d) N_2
Ans.(b)

Q45. Which type of chemical reaction involves the exchange of ions between the reactants?
(a) decomposition reaction
(b) double displacement reaction
(c) displacement reaction
(d) combination reaction
Ans.(b)

Q46. Dry ice is made by liquefying
(a) oxygen
(b) carbon dioxide
(c) carbon monoxide
(d) nitrogen
Ans.(b)

Q47. Consider the following statements:
A) A gas that follows Boyle's law, Charles' law and Avogadro law is called an ideal gas.
B) As per Gay Lussac's law, at constant volume, pressure of a fixed amount of a gas varies directly with the temperature.
C) Silicon is present in group 14 and period 3 of the periodic table.
(a) Both A and C is true
(b) Both A and B is true
(c) Only B and C is true
(d) A, B, C are true
Ans.(d)

Q48. Which of the following is not a typical property of non-metals?
(a) High ionization energies
(b) Low electronegativity
(c) Poor conductors of heat and electricity
(d) Brittle in solid state
Ans.(b)

Q49. Which of the following is the most abundant mineral in our body?
(a) Phosphorus
(b) Calcium
(c) Iron
(d) Copper
Ans.(b)

Q50. Which of the following is a non-essential amino acid as it can be made by the human body?
(a) Proline
(b) Methionine
(c) Threonine
(d) Valine
Ans.(a)

Q51. What is the vitamin deficiency in milk?
(a) D
(b) C
(c) A
(d) B Complex
Ans.(b)

Q52. Which is a major essential lipophilic (fat-soluble) vitamin required for the protection of cell membranes and the formation of red blood cells (RBCs)?
(a) Vitamin A
(b) Vitamin C
(c) Vitamin D
(d) Vitamin E

Ans.(d)

Q53. Who described the algae Spirogyra in 1674 and named the motile organismsAnimalcules, meaning 'little animals'?
(a) Maurice Wilkins
(b) Anton Van Leeuwenhoek
(c) Robert Remak
(d) BarthelemyDumortier
Ans.(b)

Q54. Who discovered a cell?
(a) Rudolf Virchow
(b) Theodor Schwann
(c) Matthias Schleiden
(d) Robert Hooke
Ans.(d)

Q55. The term plasma membrane was given by
(a) Robertson
(b) Nageli
(c) N. Grew
(d) J. Q. Plowe
Ans.(d)

Q56. Restriction enzymes were discovered by
(a) Smith and Nathans
(b) Alexander Fleming
(c) Berg
(d) None of them
Ans.(a)

Q57. Which glands help in digestion in the stomach?
(a) Pineal
(b) Gastric glands
(c) Thyroid
(d) Pituitary
Ans.(b)

Q58. Estrogen and progesterone control and stimulate the growth of?
(a) Pituitary gland
(b) Thyroid gland
(c) Mammary gland
(d) Suprarenal gland
Ans.(c)

Q59. Which nitrogenous waste produced in the body is removed by the human kidney?
(a) Only ammonia
(b) Ammonia and uric acid
(c) Urea and ammonia
(d) Urea and uric acid
Ans.(d)

Q60. Normally, in a healthy adult, the initial filtrate in the kidneys is about _____ daily.
(a) 130 L
(b) 150 L
(c) 160 L
(d) 180 L
Ans.(d)

Q61. Which among the following is a flat bone?
(a) Wrist bones
(b) Pelvic bones
(c) Ankle bones
(d) Ribs
Ans.(d)

Q62. When do humans use more facial muscles?
(a) While smiling
(b) While frowning
(c) While sleeping
(d) While talking
Ans.(b)

Q63. Which type of muscles do the uterus, iris of the eye, and bronchi contain?
(a) Striated muscles
(b) Cardiac muscles
(c) Smooth muscles
(d) Skeletal muscles
Ans.(c)

Q64. Knock-Knee syndrome results due to pollution of
(a) Heavy metal
(b) Phosphate
(c) Fluorides
(d) Nitrate
Ans.(c)

Q65. The hormone used as a oral contraceptive is .
(a) Cortisone
(b) Progesterone
(c) Testosterone
(d) Aldosterone
Ans.(b)

Q66. Malarial parasite shows which type of asexual reproduction?
(a) Binary fission
(b) Regeneration
(c) Budding
(d) Multiple fission
Ans.(d)

Q67. _____ is a typically one celled, reproductive unit capable of giving rise to a new individual without sexual fusion.
(a) Egg
(b) Spore
(c) Sperm
(d) Seed
Ans.(b)

Q68. Which of the following is NOT an example of asexual reproduction?
(a) Grafting
(b) Fragmentation
(c) Binary Fission
(d) Budding
Ans.(a)

Q69. Which among the following does not have a cell wall?
(a) Euglena
(b) Paramecium
(c) Gonyaulax
(d) Mycoplasma
Ans.(d)

Q70. The cell uses internal compartmentalization to manage its complexity. Which of the following best reflects this idea?
(a) Organelles in eukaryotic cells float freely without any specific role.
(b) Prokaryotic cells use multiple membranes to isolate cell activities.
(c) Eukaryotic cells have membrane-bound organelles that separate different functions.
(d) All cellular functions occur in the same space without any separation.
Ans.(c)

Q71. Which cell organelle is defined as the small round organelle that undergoes oxidation reaction to produce hydrogen peroxide?
(a) Centrosome
(b) Vacuole
(c) Nucleus
(d) Peroxisomes
Ans.(d)

Q72. Which of the following best explains why uracil is used in RNA instead of thymine?
(a) Uracil enhances RNA splicing.
(b) Thymine is not recognized by ribosomes.
(c) Thymine inhibits transcription.
(d) Uracil is energetically cheaper to synthesise.
Ans.(d)

Q73. What are the main energy sources for earth's internal heat engine?
(a) Radiogenic heat and oceanic tide heat
(b) Heat from volcanoes and solar heat
(c) Solar heat and oceanic tide heat
(d) Radiogenic heat and primordial heat
Ans.(d)

Q74. At what temperature do Celsius and Fahrenheit scales have the same value?
(a) 40°C
(b) 32°C
(c) -40°C
(d) -32°C
Ans.(c)

Q75. For which substance among the following, conductivity increases with temperature?
(a) Copper
(b) Germanium
(c) Silver
(d) Iron
Ans.(b)

Q76. Heat is transmitted from higher temperature to lower temperature through the actual motion of the molecules in -
(a) Conduction
(b) Convection
(c) Radiation
(d) Both conduction and convection

Ans.(b)

Q77. Cryogenics is a branch of Physics that deal with _____.
(a) very high temperatures
(b) very low temperatures
(c) growth of extremely small crystals
(d) growth of extremely large crystals
Ans.(b)

Q78. A gas occupies 2 L at 300 K and constant pressure. If temperature is increased to 450 K at constant pressure, the new volume will be
(a) 2.5 L
(b) 2.8 L
(c) 4 L
(d) 3 L
Ans.(d)

Q79. Which radiation is emitted when a stream of accelerating electrons strikes a metal target?
(a) X-ray
(b) γ-ray
(c) Cathode ray
(d) α-ray
Ans.(a)

Q80. The Rutherford nuclear model of atom predicts that atoms are unstable because the accelerated electrons revolving around the nucleus must be _____ in the nucleus.
(a) spiral
(b) circular
(c) linear
(d) elliptical
Ans.(a)

Q81. An isotope of which of the following is used in the treatment of cancer?
(a) Nickel
(b) Iron
(c) Aluminum
(d) Cobalt
Ans.(d)

Q82. What is the preferred unit used in the scientific literature to express the ratio between two sound intensities?
(a) Decibels
(b) Pascal
(c) Phon
(d) Hertz
Ans.(a)

Q83. When two sound waves with slightly different frequencies interfere, it results in a phenomenon called:
(a) Beats
(b) Harmonics
(c) Resonance
(d) Doppler effect
Ans.(a)

Q84. A sound wave travels from air into water. Which of the following characteristics of the wave will change?
(a) Frequency
(b) Wavelength
(c) Amplitude
(d) Both (a) and (b)
Ans.(b)

Q85. What is the relationship between atm and bar?
(a) $1\ atm = 10^5\ bar$
(b) $1\ atm = 1.013 * 10^5\ bar$
(c) $1\ atm = 1\ bar$
(d) $1\ atm = 1.013\ bar$
Ans.(d)

Q86. The second law of thermodynamics implies:
(a) whole of the heat can be converted into mechanical energy.
(b) no heat engine can have 100% efficiency.
(c) same heat engines working in reversible process can have 100% efficiency.
(d) a refrigerator can reduce the temperature to absolute zero.
Ans.(b)

Q87. Identify the incorrect statements.
(a) The calorific value of CNG is more than that of diesel
(b) The calorific value of kerosene is more than that of petrol
(c) The calorific value of coal is more than that of wood
(d) The calorific value of biogas is more than that of LPG
Ans.(d)

Q88. The radius of curvature of a given spherical mirror is -20 cm. The focal length of the mirror is:
(a) 40 cm
(b) -40 cm
(c) -10 cm
(d) 10 cm
Ans.(c)

Q89. Which of the following statements regarding the lenses is/are correct?
I. Power of a convex lens is represented as a positive number.
II. Power of a concave lens is represented as a negative number.
(a) Neither I nor II
(b) Both I and II
(c) Only II
(d) Only I
Ans.(b)

Q90. A fine beam of light becomes visible when it enters a smoke filled room due to :
(a) refraction of light
(b) dispersion of light
(c) reflection of light
(d) scattering of light
Ans.(d)

Q91. What is the relationship between electricity and magnetism?
(a) They are completely unrelated phenomena.
(b) They are independent phenomena that can occur separately.
(c) They are two aspects of the same fundamental force.
(d) They are both caused by a strong nuclear force.
Ans.(c)

Q92. The pattern of the magnetic field lines produced by a current carrying conductor:
(a) depends on the length of the conductor
(b) depends on the shape of the conductor
(c) is independent of anything related to the conductor
(d) depends on the thickness of the conductor
Ans.(b)

Q93. The magnetic field produced by a current carrying conductor decreases as the _____ increases.
(a) voltage
(b) current
(c) distance
(d) resistance
Ans.(c)

Q94. A star appears to shift its position in the sky when viewed from two different points in Earth's orbit. This phenomenon, used to measure stellar distance, is called:
(a) Doppler Shift
(b) Gravitational Lensing
(c) Stellar Parallax
(d) Aberration of Starlight
Ans.(c)

Q95. Electric press, heater, electric kettle, etc. are made of which element?
(a) Nichrome
(b) Carbon
(c) Constantan
(d) German Silver
Ans.(a)

Q96. The unit of self-inductance is _____.
(a) Cadilla
(b) Coltan
(c) Henry
(d) Ampere
Ans.(c)

Directions (97-101): Read the passage to answer the questions that follow.
Conformity, or acting the way most other people in one's social group act, often grows out of a person's desire for security and belonging in a group - usually a group of similar age, culture, religion, or educational background. Acting differently from the group carries the risk of social rejection, a deep fear that many people have. The drive to conform is often particularly powerful for adolescents, for whom acceptance by peers can be a primary goal, but it also affects people of all ages. Some studies suggest that conformity decreases with age.

Although the word often has a negative connotation, conformity is not necessarily a bad thing. The tendency of people to conform helps society to function smoothly in many ways. Following rules for driving, for example, enables safe transportation. The tendency of people to follow norms of interpersonal interaction enables effective communication and collaboration.

Psychologist Herbert Kelman identified and labeled three major types of conformity. The first, compliance, occurs when a person conforms publicly, but privately keeps his or her own original beliefs. People comply because of a need for approval from others and fear of rejection. The second, identification, is conforming to a particular person who is well liked and respected, such as a friend or a family member. Identification is usually motivated by the perceived role model's attractiveness or success. Internalization is when people have actually internalized a group's belief system and see it as truly their own, both publicly and privately. This is the most profound form of conformity and is likely to stay with people for a long time.

The strong force of conformity is well documented in psychological research. A researcher named Muzafer Sherif, for example, wanted to know how many people would change their opinions about something because of the desire to conform to a larger group. He conducted an experiment in which people were positioned in a dark room and asked to stare at a small dot of light 15 feet away. They were then asked to guess the amount by which the light moved (the dot actually was not moving at all). On the first day, each person saw different degrees of movement; but from the second to the fourth day, the same estimate was agreed on and the members of the group conformed their opinions to this estimate. Sherif believed this experiment demonstrated the way a norm develops in a society, and how people tend to fit their personal beliefs to that norm. Few people desire to be an "outlier" whose opinion differs widely from what is typical in the group.

The "dark side" of conformity has often been explored as well. The need to be accepted and the fear of punishment for deviating from group norms can in many cases impel people to unthinkingly adopt negative attitudes or behaviors. A phenomenon called the "spiral of silence" can occur when one opinion becomes entirely socially dominant because those people who have different opinions are afraid to speak or act on them. History is rife with examples of mass conformity to group norms and ideas that were destructive.

Q97. Which of these is NOT true about 'conformity'?
(a) behaving like others in the group
(b) need to be part of the group
(c) increases the risk of social isolation
(d) fulfils the desire for security
Ans.(c)

Q98. In what context is 'rules for driving' referred to?
(a) as an advantage of conformity
(b) as a danger of rash driving
(c) as an example of feeling of security
(d) as a primary goal of conformity
Ans.(a)

Q99. The most intense form of conformity is conformity arising out of:
(a) peer pressure
(b) identifying with a well- known figure
(c) fear of rejection
(d) holding on to one's beliefs
Ans.(d)

Q100. Sherif's experiment shows how:
(a) a norm gets established in society
(b) people continue to have faith in their beliefs
(c) strong people do not conform to society
(d) people react differently to the same situation
Ans.(a)

Q101. 'spiral of silence' means:
(a) when all in a group become silent
(b) when one opinion is allowed to dominate in society
(c) when a group develops negative attitudes
(d) examples from history about group conformity
Ans.(b)

Q102. Choose the option that is one word substitution for the given phrase.
Inscription on a tombstone
(a) epigram
(b) epic
(c) epitaph
(d) epistle
Ans.(c)

Q103. Choose the option that is the correct antonym for the given word.
PRISTINE
(a) earliest
(b) spotless
(c) dirty
(d) pure
Ans.(c)

Q104. Choose the correct option to fill in the blanks.
_____ spectators continued to watch the game despite the rain.
(a) The few
(b) Several
(c) Any
(d) Every
Ans.(b)

Q105. Choose the option that is the correct order of this jumbled sentence.
be careful/the footprints/we must/not to disturb
(a) Careful not to disturb the footprints we must be.
(b) We must be careful the footprints not to disturb
(c) The footprints we must not to disturb be careful.
(d) We must be careful not to disturb the footprints.
Ans.(d)

Q106. Choose the correct option to fill in the blanks.
My friend went to _____ university in America and got _____ MBA degree.
(a) an, a
(b) a, an
(c) the, a
(d) the, an
Ans.(b)

Q107. Choose the option that is the correct synonym for the given word.
RUSTIC
(a) artless
(b) cultured
(c) urban
(d) sophisticated
Ans.(a)

Q108. Choose the option which is the correct passive form of the given sentence.
Why did no one question her for her bad conduct?
(a) Why is she not questioned by anyone for her bad conduct?
(b) Why was she not questioned by anyone for her bad conduct?
(c) Why is she not being questioned by anyone for her bad conduct?
(d) Why has she not been questioned by anyone for her bad conduct?
Ans.(b)

Q109. Select the option that will improve the highlighted part of the sentence. In case there is no improvement needed then select 'No improvement' option.
No opinion poll can predict the overview of this election.
(a) inclination
(b) outcome
(c) estimation
(d) no improvement
Ans.(b)

Q110. Choose the correct option to fill in the blanks.
A violation of law _____ small, is a punishable offense.
(a) and
(b) nevertheless
(c) however
(d) so
Ans.(c)

Q111. Select the option that will improve the highlighted part of the sentence. In case there is no improvement needed then select 'No improvement' option.
I went down from 80 kgs to 60 kgs in weight. That is a large accomplishment.
(a) suitable accomplishment
(b) big accomplishment
(c) heavy accomplishment
(d) no improvement
Ans.(b)

Q112. Choose the option that conveys the meaning of the highlighted idiom.
We all wanted to keep the party a secret, but my brother spilled the beans.
(a) spoil the party
(b) scatter the beans on the floor
(c) reveal information
(d) not being a part of something
Ans.(c)

Q113. Choose the option which is the correct indirect form of the given sentence.
Gayatri said to her father, "Please let me go out with my friends."
(a) Gayatri requested her father to let her go out with her friends.
(b) Gayatri requested her father to let him go out with her friends.
(c) Gayatri requested her father to let him go out with his friends.
(d) Gayatri questioned her father to let her go out with her friends.
Ans.(a)

Q114. Select the most appropriate synonym of the highlighted word in the given sentence.
Simi could barely carry the pail of water as it was very heavy.
(a) Cannister
(b) Glass
(c) Bucket
(d) Bowl
Ans.(c)

Q115. Select the INCORRECTLY spelt word.
(a) Rememberance
(b) Utterance
(c) Leisure
(d) Grievance
Ans.(a)

Q116. Select the INCORRECTLY spelt word in the given sentence.
The fermentation phenomenan was formerly thought to be weird and mysterious.
(a) mysterious
(b) weird
(c) phenomenan
(d) fermentation
Ans.(c)

Q117. Select the grammatically correct form of the given sentence from the following options.
They started the cricket match therefore the rain had stopped.
(a) They started the cricket match as soon as the rain had stopped.
(b) They started the cricket match as the rain had stopped.
(c) They started the cricket match even then the rain had stopped.
(d) They started the cricket match provided that the rain had stopped.
Ans.(a)

Q118. Sentences of a paragraph are given below in jumbled order. Arrange the sentences in the correct order to form a meaningful and coherent paragraph.
A. When caffeine is ingested, it quickly enters the brain, where it competes with a chemical called adenosine.
B. One of adenosine's most important jobs is to make you feel tired.
C. But that doesn't explain why about 85 per cent of Americans consume it in some form each day.
D. Caffeine evolved in certain plants as a naturally occurring pesticide to discourage insects from eating them.
(a) DCAB
(b) DBCA
(c) BDCA
(d) ACBD
Ans.(a)

Q119. Weather is changing fast. The highlighted word is a:
(a) Noun
(b) Adjective
(c) Verb
(d) Adverb
Ans.(d)

Q120. Select the most appropriate punctuation mark to be used in place of the blank. If no punctuation mark is required, select 'No punctuation required'.
The house at the far end of the village is her_____ s.
(a) Semi colon (;)
(b) Colon (:)
(c) No punctuation required
(d) Apostrophe (')
Ans.(c)