



KTET Category-3 Part-3

Previous Year Paper
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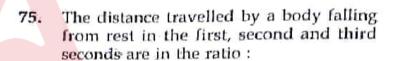
CATEGORY - III

PART - 3

Physical Science Question Numbers 71 to 150

- In longitudinal wave the direction of 71. Which among the following is an ore of 74. zinc metal?
 - (A) Bauxite
 - Calamine (B)
 - (C) Dolomite
 - Spodumene (D)

- vibration of particles of the medium is :
 - Perpendicular to the propagation of wave
- Parallel to the propagation of wave
 - Different from each other (C)
 - Varies with time to time (D)
- For pure water and perfectly clean glass, 72. the angle of contact is:
 - 90° (A)
 - (B) 45°
 - 80 (C)
 - (D)

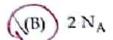


- 1:2:3
- (B) 1:3:5
 - (C) 1:4:9
 - (D) 1:5:7
- Device used to measure potential 73. difference:
 - (A) Ammeter
 - (B) Wattmeter
 - Voltmeter (C)
 - (D) Voltameter

- The focal length of lens in a camera is 76. 5 cm. What is the power of the lens?
 - (A) +0.2D
 - (B) +5D
 - (C) +10D
 - (D) +20D

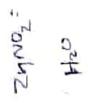
77. The number of nitrate ions in one mole of zinc nitrate is (N_A is the Avogadro Number):



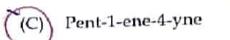








- 80. Which among the following compounds contain the maximum number of pi-bonds?
 - (A) Acetylene 📬
 - (B) Hexyne



- (D) 2-Methyl-1, 3-butadiene
- 78. The number of moles of chlorine molecules produced when 2 moles of potassium permanganate reacts with 16 moles of concentrated hydrochloric acid is:
 - (A) 2
 - (B) 3
 - (C) 5
 - (D) 8

- 81. A lens immersed in liquid become invisible when:
 - (A) the focal length of lens is zero
 - (B) the focal length of lens is infinity
 - (C) the refractive index of lens is less than that of refractive index of liquid
 - (D) the refractive index of lens is same as that of refractive index of liquid
- 79. Amount of work done in bringing unit positive charge from infinity to a point in electric field is called:
 - (A) Coulomb force
 - (B) ∠Electrostatic force
 - ((C)) Electric potential
 - (D) Hesistance

- 82. Haber process is associated with the industrial preparation of :
 - (A) Acetylene
 - (B) Ammonia
 - (C) Nitric acid
 - (D) Sulphuric acid

- 83. Which of the following is not a transverse wave?
 - (A) Sound
 - (B) Visible light
 - (C) Infrared
 - (D) Radio wave
- 84. Which of the following is not a unit of time?
 - (A) leap year
 - (B) lunar month
 - (C) light year
 - (D) micro second
- 85. Photoelectric effect was best explained by :
 - (A) Albert Einstein
 - (B) Heinrich Hertz
 - (C) Isaac Newton
 - (D) Max Planck
- 86. The acid having the highest basicity among the following is:
 - (A) Acetic acid
 - (B) Nitric acid \$
 - (C) Oxalic acid -
 - (D) Phosphoric acid

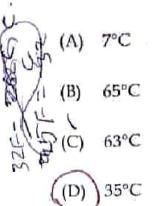
- 87. The correct electronic configuration of potassium ion in K₂SO₄ is:
 - (A) 1s²2s²2p⁶3s²3p⁶
 - (B) 1s²2s²2p⁶3s²3p⁶4s¹
 - (C) 1s²2s²2p⁶3s²3p⁶4s²
 - (D) 1s²2s²2p⁶3s²3p⁶4s²3d¹
- 88. Which among the following is not usually a property of an ionic compound?
 - (A) Conducts electricity in the solid state
 - (B) Crystalline nature
 - (C) Dissolves in polar solvents
 - (D) High melting points
- 89. Which among the following is FALSE regarding the reaction given below?

$$2A1 + Fe_2O_3 \rightarrow Al_2O_3 + 2Fe$$

- (A) Iron obtained in this reaction is usually in the molten state
- (B) | It is known as thermite reaction
- (C) The reaction is exothermic in nature
- (D) The reaction shows that aluminium is less reactive than iron

- 90. The second most electronegative element among the following is:
 - (A) Chlorine
 - *(B) Fluorine
 - (C) Nitrogen
 - (D) Oxygen
- 91. Which among the following is not a macromolecular colloid?
 - (A) colloid of protein
 - (B) colloid of rubber
 - (C) soap solution
 - (D) gold sol
 - 92. Unit of Pressure is:
 - (7A) N/m2
 - (B) Newton
 - (C) kg m/s²
 - (D) Joule
 - 93. Which among the following does not conduct electricity?
 - (A) A mixture of ammonia and vinegar in water
 - (B) Acidified water
 - (C) Water containing dissolved sugar
 - (D) Water containing dissolved NaCl

94. Temperature in Celsius corresponding to 95°F:



- 95. A liquid mixture which distils over at a constant temperature without any change in composition of the mixture is called:
 - (A) √a solvent
 - (B)

 √an allotrope
 - (C) an azeotrope
 - (D) ≠an elutant
- 96. Solar eclipse takes place when :
 - (A) the sun is between the moon and earth
 - (B) the earth is between the moon and sun
 - (C) the moon is between the sun and
 - (D) the moon does not lie on the line joining the sun and earth

- 97. Frequency of sound wave having wavelength 10 cm in air is, (Given velocity of sound = 340 m/s)
 - (A) 34 Hertz
 - (B) 340 Hertz
 - (C) 3.4 kilo Hertz
 - (D) 34 mega Hertz
- 98. Which among the following pairs of compounds have the metals in the same oxidation state?
 - (A) Ca(NO₃)₂ and AlCl₃
 - (B) Fe₂O₃ and CrCl₃.
 - (C) MnCl₂ and Na₂O
 - (D) MnO₂ and Ca₃P₂
- 99. Ultrapure germanium used in the manufacture of semiconductors is refined by:
 - ((A) Zone refining
 - (B) 'Monds process
 - (C) Van Arkel process
 - (D) Clarkes process

- 100. Which among the following is a heterogeneous mixture?
 - (A) Iron filings + Sulphur
 - (B) Oxygen + Nitrogen →
 - (C) Petrol + Alcohol
 - (D) Water + Alcohol
- Water reaches its highest density at a temperature of _______ degrees Celsius.
 - (A) 0
 - (B) 4-
 - (C) -10
 - (D) -4
- 102. Carboxylic acids are isomeric with:
 - (A) alcohols
 - (B) ethers
 - (C) esters
 - (D) aldehydes
- 103. Bubbles produced in an aquarium increase in size as it rises upward. This observation is in accordance to:
 - (A) Avogadro's law
 - (B) Boyle's law
 - (C) Charle's law
 - (D) Graham's law



104.	Size of image is always smaller than object
	in:

- (A) Convex mirror
 - (B) Concave mirror
 - (C) Plane mirror
 - (D) None of the above

105. Main constituent of LPG:

- (A) Methane
- (B) Ethane
- (C) Propane
- (D) Butane

106. If a force acts perpendicular to the direction of motion of a body, the amount of work done is:

- (A) Infinity
- (B) Constant
- (C) Zero
 - (D) Negative
- 107. If the current in a wire is 8 amperes, how much charge will flow through it in 2 seconds?
 - (A) 16 coulombs
 - (B) 8 coulombs
 - (C) 6 coulombs
 - (D) 2 coulombs

108. Final stage of evolution of a sun like star;

- (A) Black dwarf
 - (B) White dwarf
 - (C) Neutron star
- (D) Black hole

109. Which of the following is not a primary colour?

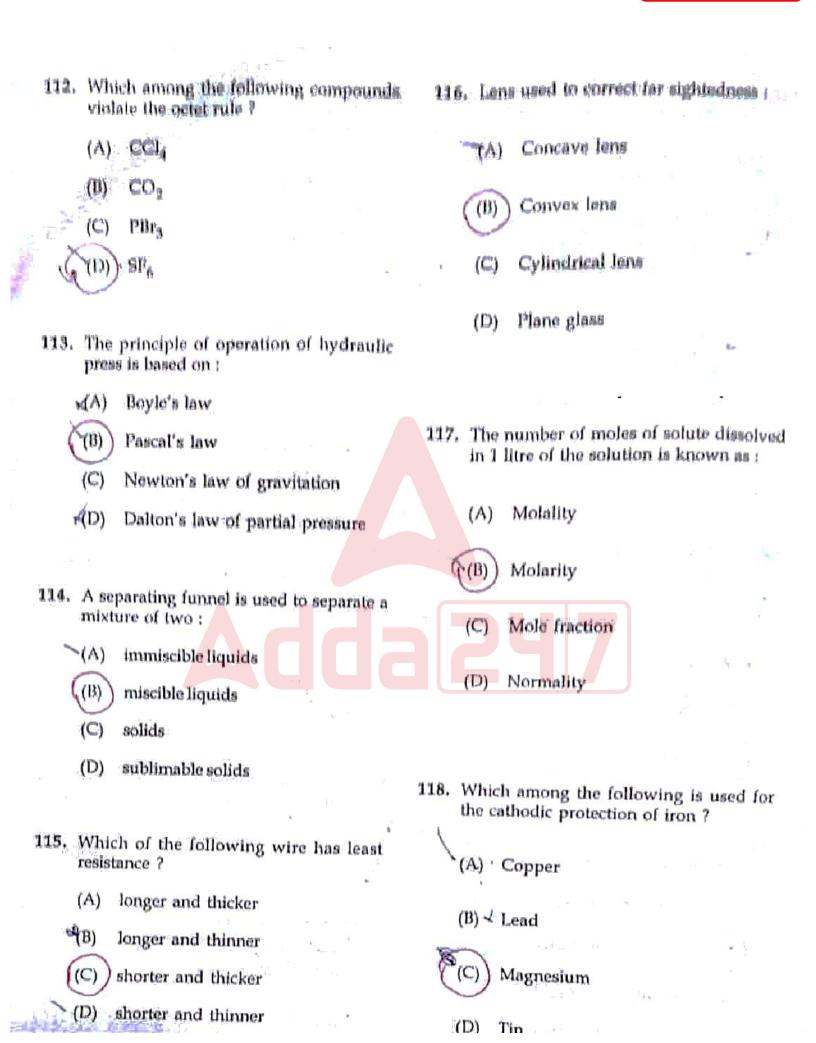
- (A) Blue
 - (B) Green
 - (C) Red
- (D) Yellow

110. When a body is taken from the equator to the earth's north pole, its:

- (A) mass and the weight remains the
- (B) mass increases weight remains the
- weight increases mass remains the
 - (D) mass and weight decreases

111. Number of valence electrons in a silicon atom is:

- (A) 5
- (B) 4
 - (C) 3
 - (D) 2



- true regarding surface tension?
 - Capillary action is a property of surface tension.
 - In the case of mercury taken in a glass tube, the cohesive force is lesser than adhesive force.
 - strong that Liquids (C) intermolecular forces also have high surface tensions.
 - Surface tension is the amount of (D) energy required to stretch or increase the surface of a liquid by a unit area.
- 120. The amphoteric oxide among the following is:
 - CaO
 - CO_2
 - NO2
 - ZnO
- 121. Structural and hierarchical arrangement of concept describes in:
 - Concept Attainment Model (A)
 - Advance Organiser Model
 - (C) Inquiry Training Model
 - (D) Group Investigation Model

- 119. Which of the following statements is NOT 122. Who established Tata Institute of Fundamental Research Mumbai ?
 - Bhabha $\langle A \rangle$
 - Einstein' (B)
 - Faraday (C)
 - Max Plank (D)
 - 123. Which of the following is exclusively meant for learner?
 - Reference book (A)
 - Hand book (B)
 - (C) Source book
 - Work book (D)
 - 124. According to Bruner, the stage at which knowledge is represented through images is called:
 - (A) Enactive stage
 - Iconic stage
 - Symbolic stage
 - (D) Visual stage



125. Which one of the following does not promote reflection ?	129. National Curricular Framework (NCF 2005) lays more emphasize on :
(A) Instrumental feedback	(A) Behaviorism
(B) Case records	(B) Naturalism
(C) Journal	(C) Pragmatism
(D) Hand book	(D) Constructivism
126. The professional growth of science teachers can be ensured maximum by providing;	130. Which among the following is not a Herbartian approach to lesson planning?
(A) In-service training	(A) Association
(B) Source book	(B) Generalization
(C) Question bank	(C) Application
(D) Attractive salary	(D) Induction
127. The evaluation at the end of the course is: (A) Placement evaluation (B) Formative evaluation (C) Diagnostic evaluation (D) Summative evaluation	131. The term scaffolding is related to : (A) Piaget (B) Bruner (C) Vygotsky (D) Skinner
128. The most appropriate chart to show the difference between nuclear fission and nuclear fusion:	132. The approach based on the principle 'general to specific' in organizing curriculum is:
(A) Flow chart	(A) Concentric approach
(B) Tabular chart	(B) Unit approach
(C) ∀Content chart	(C) Topic approach
(A(D)) Tree chart	(D) Spiral approach

- 133. Before teaching the topic gravitation the teacher explains the years of hard-work of Sir Isaac Newton which resulted in the discovery of the concept. The teacher is using:
 - (A) Scientific Method
 - (B) Problem Solving Method
 - +(C) Discovery Method
 - (D) Historical Method
- 134. Identify the most suitable technique for transacting the concept of Rutherford's Scattering Experiment:
 - (A) Brainstorming
 - (B) Buzz session
 - (C) Dramatization
 - (D) Computer simulation
- 135. The logical and systematic breakup of the curriculum from the point of view of a pedagogue is:
 - (A) Technological analysis
 - (B) Pedagogical analysis
 - (C) Content analysis
 - (D) Techno pedagogic analysis

- 136. Which of the following is not a characteristic of scientific attitude?
 - (A) Open mindedness
 - (B) Critical thinking
 - (C) Objectivity
 - (D) Easily accepting others opinion
- 137. While teaching the concept 'force', the size and shape of objects are considered as :
 - (A) Essential-Attributes
 - (B) Non-exemplars
 - (C) Non-essential attributes
 - (D) Exemplars
- 138. An instance where knowledge of Mathematics is NOT inevitable to understand Physics is:
 - (A) Arriving at the concept of Boyle's law
 - (B) Illustrating laws of motion
 - (C) Classifying simple machines to levers and non-levers
 - (D) Establishing relation between work and energy

- 139. Identify the test used to find out area of difficulty faced by students:
 - (A) Achievement Test
 - (B) Diagnostic Test
 - (C) Prognostic Test
 - (D) Unit Test

- 142. What type of question is useful for collecting more creative ideas regarding the harmful effects of artificial drinks?
 - (A) Multiple choice
 - (B) Sentence completion
 - (C) Short answer
 - (D) Essay
- 140. Which of the following does not come under McCormack and Yager's Taxonomy?
 - (A) Process Domain
 - (B) Creativity Domain
 - (C) Attitudinal Domain
 - (D) Affective Domain

- 143. OHP is an example for :
 - (A) Activity aid
 - (B) Projected aid
 - (C) Non Projected aid
 - (D) Chart

- 141. Cognitive domain explains:
 - (A) Doing field
 - (B) Knowledge field
 - (C) Feeling field
 - (D) Activity field

- 144. 'Science is the pursuit of truth'. The value of science referred here is:
 - (A) Moral
 - (B) Practical
 - (C) Social
 - (D) Intellectual

145.	Identify	the concept	from	the	following:	148.	- 9
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- (A) Helium is an inert gas
- (B) Iron is a hard metal
 - (C) Sugar solution is a homogeneous mixture
- (D) Metals conduct electricity
- 146. Who among the four educationalists is closely associated with project method?
 - (A) Kilpatrick
 - (B) Thorndike
 - (C) Bloom
 - (D) Gagne
- 147. Which among the following is not associated with Piaget?
 - (A) Schema
 - Zone of Proximal Development
 - (C) Assimilation
 - (D) Accommodation

148. Which among the following is an activity aid?

- (A) Video clips
- (B) Working model
 - (C) Field trip
 - (D) Still model

149. Which of the following is not the property of CCE?

- (A) Diagnosing weaknesses of students
- (B) Informal evaluation carried out by teachers
- (C) Multiple techniques of evaluation
- (D) Considering only scholastic aspects

150. If there is purchase of some non-consumable articles in the science laboratory, it should be entered in:

- (A) Permanent stock register
 - (B) Breakable register
 - (C) Order register
 - (D) Requirement register